

## Chemical Safety Data Sheet MSDS / SDS

## Propionic anhydride

Revision Date:2026-05-31 Revision Number:1

## SECTION 1: Identification of the substance/mixture and of the company/undertaking

## Product identifier

Product name : Propionic anhydride  
CBnumber : CB7852974  
CAS : 123-62-6  
EINECS Number : 204-638-2  
Synonyms : PROPIONIC ANHYDRIDE,propionic

## Relevant identified uses of the substance or mixture and uses advised against

Relevant identified uses : For R&D use only. Not for medicinal, household or other use.  
Uses advised against : none

## Company Identification

Company : Chemicalbook  
Address : Building 1, Huihuang International, Shangdi 10th Street, Haidian District, Beijing  
Telephone : 010-86108875

## SECTION 2: Hazards identification

## Classification of the substance or mixture

Skin corrosion, Sub-category 1B

## Label elements

## Pictogram(s)

□

Signal word : Danger

## Hazard statement(s)

H227 Combustible liquid

H314 Causes severe skin burns and eye damage

## Precautionary statement(s)

P280 Wear protective gloves/protective clothing/eye protection/face protection.

P303+P361+P353 IF ON SKIN (or hair): Remove/Take off Immediately all contaminated clothing. Rinse SKIN with water/shower.

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do.

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P405 Store locked up.

### Prevention

P260 Do not breathe dust/fume/gas/mist/vapours/spray.

P264 Wash ... thoroughly after handling.

P280 Wear protective gloves/protective clothing/eye protection/face protection/hearing protection/...

### Response

P301+P330+P331 IF SWALLOWED: Rinse mouth. Do NOT induce vomiting.

P363 Wash contaminated clothing before reuse.

P304+P340 IF INHALED: Remove person to fresh air and keep comfortable for breathing.

P316 Get emergency medical help immediately.

P321 Specific treatment (see ... on this label).

P305+P351+P338 IF IN EYES: Rinse cautiously with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing.

### Storage

P405 Store locked up.

### Disposal

P501 Dispose of contents/container to an appropriate treatment and disposal facility in accordance with applicable laws and regulations, and product characteristics at time of disposal.

### Other hazards

no data available

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## SECTION 3: Composition/information on ingredients

### Substance

|              |                                 |
|--------------|---------------------------------|
| Product name | : Propionic anhydride           |
| Synonyms     | : PROPIONIC ANHYDRIDE,propionic |
| CAS          | : 123-62-6                      |
| EC number    | : 204-638-2                     |
| MF           | : C6H10O3                       |
| MW           | : 130.14                        |

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## SECTION 4: First aid measures

### Description of first aid measures

#### If inhaled

Fresh air, rest. Artificial respiration may be needed. Refer for medical attention.

#### Following skin contact

Remove contaminated clothes. Rinse skin with plenty of water or shower. Refer for medical attention .

#### Following eye contact

First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then refer for medical attention.

#### Following ingestion

Rinse mouth. Do NOT induce vomiting. Give one or two glasses of water to drink. Refer for medical attention .

### Most important symptoms and effects, both acute and delayed

Inhalation causes irritation of eyes and respiratory tract. Contact with liquid causes burns of eyes and skin. Ingestion causes burns of mouth and stomach. (USCG, 1999)

### Indication of any immediate medical attention and special treatment needed

First aid: Inhalation exposure: Fresh air, rest. Artificial respiration if indicated. Refer for medical attention. Skin exposure: Remove contaminated clothes. Rinse skin with plenty of water or shower. Eye exposure: First rinse with plenty of water for several minutes (remove contact lenses if easily possible), then take to a doctor. Ingestion exposure: Rinse mouth. Do NOT induce vomiting. Give plenty of water to drink. Refer for medical attention. from table

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## SECTION 5: Firefighting measures

### Extinguishing media

Use water spray, dry chemical, "alcohol resistant" foam, or carbon dioxide. Use water spray to keep fire-exposed containers cool. Hazardous reactions may be avoided by using large quantities of water to solubilize the anhydride & fully absorb the heat of reaction.

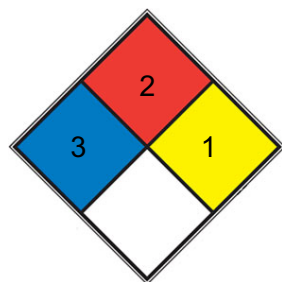
### Specific Hazards Arising from the Chemical

Excerpt from ERG Guide 156 [Substances - Toxic and/or Corrosive (Combustible / Water-Sensitive)]: Combustible material: may burn but does not ignite readily. Substance will react with water (some violently) releasing flammable, toxic or corrosive gases and runoff. When heated, vapors may form explosive mixtures with air: indoors, outdoors and sewers explosion hazards. Most vapors are heavier than air. They will spread along ground and collect in low or confined areas (sewers, basements, tanks). Vapors may travel to source of ignition and flash back. Contact with metals may evolve flammable hydrogen gas. Containers may explode when heated or if contaminated with water. (ERG, 2016)

### Advice for firefighters

Use water spray, powder, alcohol-resistant foam, carbon dioxide.

### NFPA 704



■ HEALTH 3 Short exposure could cause serious temporary or moderate residual injury (e.g. [liquid hydrogen](#), [sulfuric acid](#), [calcium hypochlorite](#), hexafluorosilicic acid)

■ FIRE 2 Must be moderately heated or exposed to relatively high ambient temperature before ignition can occur and multiple finely divided suspended solids that do not require heating before ignition can occur. Flash point between 37.8 and 93.3 °C (100 and 200 °F). (e.g. diesel fuel, [sulfur](#))

■ REACT 1 Normally stable, but can become unstable at elevated temperatures and pressures (e.g. [propene](#))

□ SPEC.

## SECTION 6: Accidental release measures

### Personal precautions, protective equipment and emergency procedures

Personal protection: chemical protection suit and face shield. Collect leaking liquid in sealable containers. Absorb remaining liquid in dry sand or inert absorbent. Then store and dispose of according to local regulations.

### Environmental precautions

Personal protection: chemical protection suit and face shield. Collect leaking liquid in sealable containers. Absorb remaining liquid in dry sand or inert absorbent. Then store and dispose of according to local regulations.

### Methods and materials for containment and cleaning up

Spill or leak procedures: Use water spray to cool and disperse vapors, protect personnel, and dilute spills to form nonflammable mixtures. Control runoff and isolate discharged material for proper disposal. Neutralize spill and washings with soda ash or lime.

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## SECTION 7: Handling and storage

### Precautions for safe handling

NO open flames. Above 63°C use a closed system and ventilation. Handling in a well ventilated place. Wear suitable protective clothing. Avoid contact with skin and eyes. Avoid formation of dust and aerosols. Use non-sparking tools. Prevent fire caused by electrostatic discharge steam.

### Conditions for safe storage, including any incompatibilities

Dry. Ventilation along the floor. Separated from bases, oxidants and food and feedstuffs. Separated from acids, bases, oxidants, food and feedstuffs. Dry. Ventilation along the floor.

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## SECTION 8: Exposure controls/personal protection

### Control parameters

#### Occupational Exposure limit values

no data available

#### Biological limit values

no data available

### Exposure controls

Ensure adequate ventilation. Handle in accordance with good industrial hygiene and safety practice. Set up emergency exits and the risk-elimination area.

### Individual protection measures

#### Eye/face protection

Wear face shield or eye protection in combination with breathing protection.

#### **Skin protection**

Protective gloves. Protective clothing.

#### **Respiratory protection**

Use ventilation, local exhaust or breathing protection.

#### **Thermal hazards**

no data available

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## SECTION 9: Physical and chemical properties

### **Information on basic physicochemical properties**

|  |   |
|--|---|
| Physical state   | Propionic anhydride is a colorless liquid with a pungent odor. Flash point 165°F. Density 8.4 lb /gal.<br>Corrosive to metals and tissue. |
| Colour   | Colorless liquid  |
| Odour  | Very rancid odor  |
| Melting point/freezing point                             | 13°C(lit.)  |
| Boiling point or initial boiling point and boiling range | 168°C   |
| Flammability   | Combustible.  |
| Lower and upper explosion limit/flammability limit       | 1.3-9.5%(V)   |
| Flash point  | 66°C  |
| Auto-ignition temperature                                | 545° F (USCG, 1999)   |
| Decomposition temperature                                | no data available   |
| pH   | 2.5 (100g/l, H <sub>2</sub> O, 20°C)(calculated on the free acid)   |
| Kinematic viscosity                                      | 1.144 cP at 20 deg C; 0.978 cP at 30 deg C; 0.853 cP at 40 deg C; 0.7511 cP at 50 deg C   |
| Solubility   | H <sub>2</sub> O: decomposes (when in contact with water)   |
| Partition coefficient n-octanol/water                    | log Kow = 0.40 (est)  |
| Vapour pressure  | 10 mm Hg ( 57.7 °C)   |
| Density and/or relative density                          | 1.015   |
| Relative vapour density                                  | 4.5 (vs air)  |
| Particle characteristics                                 | no data available   |

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## SECTION 10: Stability and reactivity

### **Reactivity**

Decomposes on contact with water. This produces corrosive propionic acid. Reacts violently with bases and oxidants. This generates fire and explosion hazard.

### **Chemical stability**

no data available

### **Possibility of hazardous reactions**

Combustible liquid The vapour is heavier than air. PROPIONIC ANHYDRIDE reacts exothermically with water. The reactions are sometimes slow, but can become violent when local heating accelerates their rate. Acids accelerate the reaction with water. Incompatible with acids, strong oxidizing agents, alcohols, amines, and bases.

### **Conditions to avoid**

no data available

### **Incompatible materials**

The substance decomposes on contact with water producing corrosive propionic acid. Reacts violently with acids, bases and oxidants causing fire and explosion hazard.

### **Hazardous decomposition products**

When heated to decomposition it emits acrid smoke and irritating fumes.

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## **SECTION 11: Toxicological information**

### **Acute toxicity**

- Oral: LD50 Rat oral 2360 mg/kg
- Inhalation: no data available
- Dermal: no data available

### **Skin corrosion/irritation**

no data available

### **Serious eye damage/irritation**

no data available

### **Respiratory or skin sensitization**

no data available

### **Germ cell mutagenicity**

no data available

### **Carcinogenicity**

no data available

### **Reproductive toxicity**

no data available

### **STOT-single exposure**

The substance is corrosive to the eyes, skin and respiratory tract. Inhalation of the vapour or aerosol may cause lung oedema. See Notes. Corrosive on ingestion. The effects may be delayed. Medical observation is indicated.

### **STOT-repeated exposure**

no data available

### **Aspiration hazard**

No indication can be given about the rate at which a harmful concentration of this substance in the air is reached on evaporation at 20°C.

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## SECTION 12: Ecological information

### **Toxicity**

Toxicity to fish: no data available

Toxicity to daphnia and other aquatic invertebrates: no data available

Toxicity to algae: no data available

Toxicity to microorganisms: no data available

### **Persistence and degradability**

Biodegradation is not expected to be an important environmental fate process due to the rapid hydrolysis of propionic anhydride under aqueous conditions(1-2). Propionic anhydride, present at 100 mg/L, reached 83% of its theoretical BOD in 2 weeks using an activated sludge inoculum at 30 mg/L in the Japanese MITI test(3), however it is believed that this biodegradation measurement was of the acid(SRC).

### **Bioaccumulative potential**

Bioconcentration of propionic anhydride is not expected to be an important environmental fate process due to the rapid hydrolysis to propionic acid under aqueous conditions(1-2).

### **Mobility in soil**

Soil adsorption and mobility of propionic anhydride are not expected to be important environmental fate processes due to the rapid hydrolysis to propionic acid under aqueous conditions(1-2).

### **Other adverse effects**

no data available

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## SECTION 13: Disposal considerations

### **Disposal methods**

#### **Product**

The material can be disposed of by removal to a licensed chemical destruction plant or by controlled incineration with flue gas scrubbing. Do not contaminate water, foodstuffs, feed or seed by storage or disposal. Do not discharge to sewer systems.

#### **Contaminated packaging**

Containers can be triply rinsed (or equivalent) and offered for recycling or reconditioning. Alternatively, the packaging can be punctured to make it unusable for other purposes and then be disposed of in a sanitary landfill. Controlled incineration with flue gas scrubbing is possible for combustible packaging materials.

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## SECTION 14: Transport information

## **UN Number**

ADR/RID: UN2496 (For reference only, please check.)

IMDG: UN2496 (For reference only, please check.)

IATA: UN2496 (For reference only, please check.)

## **UN Proper Shipping Name**

ADR/RID: PROPIONIC ANHYDRIDE (For reference only, please check.)

IMDG: PROPIONIC ANHYDRIDE (For reference only, please check.)

IATA: PROPIONIC ANHYDRIDE (For reference only, please check.)

## **Transport hazard class(es)**

ADR/RID: 8 (For reference only, please check.)

IMDG: 8 (For reference only, please check.)

IATA: 8 (For reference only, please check.)

## **Packing group, if applicable**

ADR/RID: III (For reference only, please check.)

IMDG: III (For reference only, please check.)

IATA: III (For reference only, please check.)

## **Environmental hazards**

ADR/RID: No

IMDG: No

IATA: No

## **Special precautions for user**

no data available

## **Transport in bulk according to IMO instruments**

no data available

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## SECTION 15: Regulatory information

### **Safety, health and environmental regulations specific for the product in question**

#### **European Inventory of Existing Commercial Chemical Substances (EINECS)**

Listed.

#### **EC Inventory**

Listed.

#### **United States Toxic Substances Control Act (TSCA) Inventory**

Listed.

#### **China Catalog of Hazardous chemicals 2015**

Listed.

#### **New Zealand Inventory of Chemicals (NZIoC)**

Listed.

#### **PICCS**

Listed.

#### **Vietnam National Chemical Inventory**

Listed.

#### **IECSC**

Listed.

#### **Korea Existing Chemicals List (KECL)**

Listed.

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## SECTION 16: Other information

### **Abbreviations and acronyms**

CAS: Chemical Abstracts Service

ADR: European Agreement concerning the International Carriage of Dangerous Goods by Road

RID: Regulation concerning the International Carriage of Dangerous Goods by Rail

IMDG: International Maritime Dangerous Goods

IATA: International Air Transportation Association

TWA: Time Weighted Average

STEL: Short term exposure limit

LC50: Lethal Concentration 50%

LD50: Lethal Dose 50%

EC50: Effective Concentration 50%

### **References**

IPCS - The International Chemical Safety Cards (ICSC), website: <http://www.ilo.org/dyn/icsc/showcard.home>

HSDB - Hazardous Substances Data Bank, website: <https://toxnet.nlm.nih.gov/newtoxnet/hsdb.htm>

IARC - International Agency for Research on Cancer, website: <http://www.iarc.fr/>

eChemPortal - The Global Portal to Information on Chemical Substances by OECD, website: [http://www.echemportal.org/echemportal/index?pageID=0&request\\_locale=en](http://www.echemportal.org/echemportal/index?pageID=0&request_locale=en)

CAMEO Chemicals, website: <http://cameochemicals.noaa.gov/search/simple>

ChemIDplus, website: <http://chem.sis.nlm.nih.gov/chemidplus/chemidlite.jsp>

ERG - Emergency Response Guidebook by U.S. Department of Transportation, website: <http://www.phmsa.dot.gov/hazmat/library/erg>

Germany GESTIS-database on hazard substance, website: <http://www.dguv.de/ifa/gestis/gestis-stoffdatenbank/index-2.jsp>

ECHA - European Chemicals Agency, website: <https://echa.europa.eu/>

### **Other Information**

The symptoms of lung oedema often do not become manifest until a few hours have passed and they are aggravated by physical effort. Rest and medical observation are therefore essential. Immediate administration of an appropriate inhalation therapy by a doctor or a person authorized by him/her, should be considered.

#### **Disclaimer:**

The information in this MSDS is only applicable to the specified product, unless otherwise specified, it is not applicable to the mixture of this product and other substances. This MSDS only provides information on the safety of the product for those who have received the

appropriate professional training for the user of the product. Users of this MSDS must make independent judgments on the applicability of this SDS. The authors of this MSDS will not be held responsible for any harm caused by the use of this MSDS.